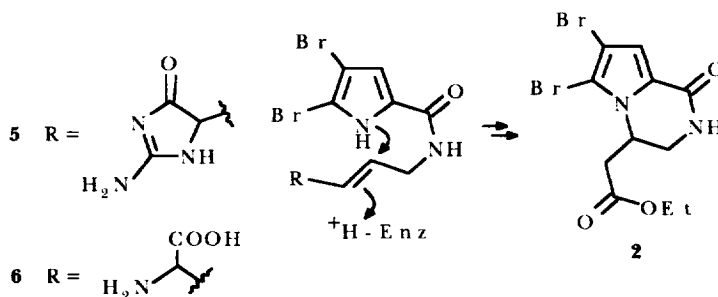


respectively; CONH₂ as two broad signals at δ_{H} 6.49 and 7.20 for NH₂ and a s at δ_{C} 158.10 for C=O; C-2 and C-4 as s at δ_{C} 127.82 and 96.57 (upfield shift by Br), respectively. The composition C₁₁H₁₂N₂O₃Br₂ for hanishin (**2**)⁷ was similarly derived while the pyrrole moiety rests on singlet signals at δ_{C} 106.15, 99.55 and 127.48 and a doublet signal at 115.17, where C-4 and C-2 assignments were made by ¹H-¹³C correlation and inverse-detection HMBC⁸ correlation of the pyrrole-fitting δ_{H} 6.86 s with δ_{C} 115.17 d and 106.15 s, respectively, while C-3 was assigned from high-field shift (δ_{C} 99.55) by bromine. The amide group was assigned from δ_{C} 158.54 s (C=O) and a δ_{H} 6.95 broad signal (H-7), the latter coupled, as indicated by selective irradiations, with H₂-8, which showed up as δ_{H} 4.00 ddd and δ_{H} 3.68 ddd signals, both correlated with δ_{C} 36.58 t. The methine group was revealed as a δ_{H} 4.82 dt coupled with δ_{C} 51.68 d as well as with both H₁-8 (δ_{H} 2.96 dd) and H₂-10 (δ_{H} 2.64 ddd). The latter was also correlated with δ_{C} 43.65 t. Accounting for the unsaturation due to the ester group - fully supported by 1D and 2D NMR data - this required N-1/C-9 ring closure. The connectivity C-9/C-10 was confirmed by HMBC⁸ correlation of H₁-10 (δ_{H} 2.96 dd) with C-8 (δ_{C} 36.58 t).

Hanishin (**2**), although giving no optical rotation across a range of wavelengths, showed a weak Cotton effect.⁷ Since no model compound was available to relate the CD elongation to the optical purity, experiments with the chiral shift reagent Eu(tfc)₃ were carried out, revealing the splitting of δ_{H} 2.94 dd (10-H) into two signals in 3:2 integration ratio, which represents also the enantiomeric excess for **2**.

Since alkaloids in this group often show therapeutically promising bioactivity¹ or have ecological relevance,^{5b,9} compounds **1-4** were screened *in vitro* against NSCLC-N6 human non-small-cell-lung carcinoma, recording IC₅₀ 11.2, 9.7, 4.8 and 9.4 $\mu\text{g/ml}$, respectively. This may be a case where weak biological activities are worthwhile pursuing,¹⁰ not only because there is little treatment for this form of tumour, but also because hanishin and analogues should prove easily amenable to total synthesis, allowing screening of both enantiomers. As it is often the case with antimicrobial agents,¹¹ the free carboxylic acid form of hanishin could prove far more cytotoxic than the ester, although this is not our criterion of choice: we are aimed at inducing terminal differentiation of these tumour cells, which would be the basis for suppressing their neoplastic character, and low-activity compounds may give hope of no general toxicity.



Scheme 1

Oroidin type alkaloids are considered condensation products of highly modified prolines, which have been separately isolated from *Agelas mauritiana*^{12a} and *Lissodendoryx* sp.,^{12b} the latter belonging to an atypical order (Poecilosclerida) for these types of alkaloids. Additions to¹³ or deviations from¹⁴ this scheme have also been

proposed for related alkaloids. The biogenesis of hanishin may be viewed, as in Scheme 1, from 1-derived aminoimidazolinone **5** or amino acid **6** intermediates *via* N(1) → C(9) (hanishin numbering) cyclization followed by oxidative breakdown of the side chain. Alternatively, breakdown of the chain may precede cyclization. Anyway, any racemization of hanishin during work up was ruled out and it can not be easily imagined for any conceivable precursor of hanishin. Thus, unlike certain cases of phenol coupling,¹⁵ it is difficult to envision formation of hanishin other than by the aid of an enzyme or dirigent protein¹⁶ system, like in the above routes. The low enantiomeric purity of hanishin may be related to the highly polymorphic condition of *A. carteri*, different collections of which gave different, though intact, C₁₁ alkaloids of the oroidin family.¹⁷ This suggests that *A. carteri* of the Hanish Islands saved only a memory of the ability of congeners^{1,12-14,17} to bring about cyclization of oroidin or related, linear precursors. Perhaps our sponge does badly this biosynthetic function as to the stereochemistry by the way of a vestigial enzyme, as for a shunt metabolite.

We thank the Ardoukoba Association and Daniel Jouvance Society for support in collecting the sponge, Dr. J. Vacelet for the sponge identification, Mrs M. Rossi and A. Sterni for technical contributions in the isolation of metabolites and mass spectra, respectively, and MURST 40% and CNR (also Progetto Strategico 96-05073), Roma, for financial support. Finally, an anonymous referee deserves our gratitude for very constructive criticism.

REFERENCES AND NOTES

- D'Ambrosio, M.; Guerriero, A.; Debitus, C.; Ribes, O.; Pusset, J.; Leroy, S.; Pietra, F. *J. Chem. Soc., Chem. Commun.* **1993**, 1305-1306; D'Ambrosio, M.; Guerriero, A.; Ripamonti, M.; Debitus, C.; Waikedre, J.; Pietra, F. *Helv. Chim. Acta* **1996**, *79*, 727-735.
- Mancini, I.; Guella, G.; Amade, P.; Pietra, F. *Tetrahedron* **1997**, *53*, 2625-2628.
- Fractions 20-22 of previous chromatographic (FC) work up of the ethanolic extracts of the sponge² were subjected to reversed-phase RP-18 HPLC with H₂O-MeCN 2:3, solvent flow 5 ml min⁻¹ under UV monitoring at λ 254 nm, to give **4** (*t_R* 4.0 min, 3.0 mg, 0.022% on raw extract), **3** (*t_R* 5.5 min, 1.2 mg, 0.009%) and **2** (*t_R* 12.0 min, 2.5 mg, 0.087%). Amine-HPLC (EtOAc-MeOH 8:2 with added Et₃N 5%, λ 280 nm) of fraction 25 from the above FC gave oroidin **1** (*t_R* 6.5 min, 4.3 mg, 0.031%).
- Garcia, E.E.; Benjamin, L.E.; Fryer, I.R. *J. Chem. Soc., Chem. Commun.* **1973**, 78-79.
- a) Tada, H.; Tozko, T. *Chem. Lett.* **1988**, 803-806; b) Tsukamoto, S.; Kato, H.; Hirota, H.; Fusetani, N. *J. Nat. Prod.* **1996**, *59*, 501-503.
- Data for **4**: semisolid; UV (MeOH) λ_{max} 267 nm (ε 16200 l mol⁻¹ cm⁻¹), 230 (9800), 202 (13800); ¹H NMR ((CD₃)₂CO, J/Hz, 299.94 MHz, Me₄Si) δ 10.95 (br.s, 1-H), 6.87 (dd, *J*_{3,1} 2.7, *J*_{3,5} 1.5, 3-H), 7.03 (dd, *J*_{5,1} 2.7, *J*_{5,3} 1.5, 5-H), 6.49 and 7.20 (two br.s, H₂NCO); ¹³C NMR ((CD₃)₂CO, 75.43 MHz) δ 127.82 (s, C-2), 122.22 (d, C-3), 96.57 (s, C-4), 112.65 (d, C-5), 158.10 (s, C=O). EI-MS *m/z* (%) 190, 188 (M⁺, 100/100), 174, 172 ([M - NH₂]⁺, 22/22), 146, 144 (16/16), 119, 117 (10/10), 109 (6), 65 (21). HR-EI-MS *m/z* 187.957±0.004, calc. for C₅H₅N₂O⁷⁹Br 187.958).
- Data for **2**: semisolid; [α]²⁵ = 0.0 at λ 589, 546 and 435 nm (MeOH, *c* 0.1 g per 100 ml); CD (MeOH) Δε (λ) -0.4 (300), 0.1 (262), -0.6 (212); UV (MeOH) λ_{max} 280 nm (ε 6500 l mol⁻¹ cm⁻¹), 230 (10000), 203 (17800); ¹H NMR ((CD₃)₂CO, J/Hz, 299.94 MHz, Me₄Si) δ 6.86 (s, 4-H), 6.95 (br.s, 7-H), 4.00 (ddd, *J*_{gem} 13.6, *J*_{8a,9} 4.3, *J*_{8a,NH} 1.5, 8-H_a), 3.68 (ddd, *J*_{gem} 13.6, *J*_{8b,NH} 5.3, *J*_{8b,10b} 1.5, 8-H_b), 4.82 (dt, *J*_{5,10a} 10.1, *J*_{5,8a} ≈ *J*_{5,10b}

3.9, 9-H), 2.96 (dd, J_{gem} 16.0, $J_{10a,9}$ 10.1, 10-H_a), 2.64 (ddd, J_{gem} 6.0, $J_{10b,9}$ 3.9, $J_{10b,8b}$ 1.5, 10-H_b), 4.25 (q, J 7.2, O-CH₂), 1.23 (t, J 7.2, CH₃); working in CDCl₃, to **2** was added 0.010M Eu(tfc)₃ in 10- μ l portion until, at 0.2 molar equivalent of Eu(tfc)₃ added, δ_{11} 2.94 dd (10-H) was split into two signals in ca. 3:2 integration ratio; this can be taken with qualitative confidence as the integration is for the same proton in the diastereomeric complexes and relaxation was amply allowed; ¹³C NMR ((CD₃)₂CO, 75.43 MHz) δ 106.15 (s, C-2), 99.55 (s, C-3), 115.17 (d, C-4), 127.48 (s, C-5), 158.54 (s, C-6), 36.58 (t, C-8), 51.68 (d, C-9), 43.65 (t, C-10), 170.65 (s, C-11), 61.50 (t, OCH₂), 14.32 (q, CH₃). EI-MS m/z (%) 382, 380, 378 (M⁺, 17/34/17), 337, 335, 333 ([M - C₂H₅O]⁺, 4/8/4), 309, 307, 305 (1/2/1), 295, 293, 291 (37/74/37), 214, 212 (9/9), 183 (3), 144, 142 (8/8); HR-EI-MS m/z 377.921 \pm 0.004, calc. for C₁₁H₁₂N₂O₃⁷⁹Br₂ 377.921); FAB-MS (H⁺, 3-nitrobenzyl alcohol) m/z (%) 383, 381, 379 ([M + H]⁺, 20/40/20).

8. Müller, L. *J. Am. Chem. Soc.* **1979**, *101*, 4481-4484.
9. Chanas, B.; Pawlik, R.J.; Lindel, T.; Fenical, W. *J. Exp. Marine Biol. Ecol.* **1997**, *208*, 185-196.
10. Copping L.G., Ed., *Crop Protection Agents from Nature. Natural Products and Analogues*; Royal Society: London, 1996.
11. Capon, R.J.; MacLeod, J.K. *Tetrahedron* **1985**, *41*, 3391-3404; He, H.-y.; Faulkner, D.J.; Lu, H.S.M.; Clardy, J. *J. Org. Chem.* **1991**, *56*, 2112-2115.
12. a) Wright, A.E.; Chiles, S.A.; Cross, S.S. *J. Nat. Prod.* **1991**, *54*, 1684-1686; b) Schmitz, F.J.; Gunasekera, S.P.; Lakshmi, V.; Tillekeratne, L.M.V. *Ibid.* **1985**, *48*, 47-53.
13. Jeménez, C.; Crews, P. *Tetrahedron Lett.* **1994**, *35*, 1375-1378.
14. Cafieri, F.; Fattorusso, E.; Mangoni, A.; Tagliatela-Scafati, O. *Tetrahedron* **1996**, *52*, 13713-13720.
15. Guerriero, A.; D'Ambrosio, M.; Traldi, P.; Pietra, F. *Naturwissenschaften* **1984**, *71*, 425-426.
16. Davin, L.B.; Wang, H.-B.; Crowell, A.L.; Bedgar, D.L.; Martin, D.M.; Sarkanen, S.; Lewis, N.G. *Science* **1997**, *275*, 362-366.
17. Cimino, G.; De Rosa, S.; De Stefano, S.; Mazzarella, L.; Puliti, R.; Sodano, G. *Tetrahedron Lett.* **1982**, *23*, 767-768; Fedoreyev, S.A.; Utkina, N.K.; Ilyin, S.G.; Reshetnyak, M.V.; Maximov, O.B. *Ibid.* **1986**, *27*, 3177-3180.

(Received in UK 12 June 1997; revised 7 July 1997; accepted 11 July 1997)